

Structural Characterization, Raman Spectroscopy and FTIR Spectroscopy Studies of Pr³⁺ Doped Tellurium Bismuth Borate Glasses

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Abstract

A set of composition-based undoped and Pr³⁺-doped Tellurium Bismuth Borate glasses 20TeO₂: (30-X)Bi₂O₃: 50B₂O₃: XPr₂O₃ (where X= 0, 0.1, 0.3, 0.5, 0.7 mol%) have been fabricated via standard melt quenching technique. All of the manufactured glasses were made of completely amorphous material, according to X-ray diffraction (XRD) patterns. The refractive index, density, optical absorption, FTIR and Raman spectroscopy have been applied to investigate the structural and optical properties. FTIR spectral measurements have been utilized to identify the network vibration of Pr₂O₃ doped Tellurium glasses. FTIR reveals symmetrical stretching vibrations of Te-O in TeO₄ as well as B-O stretching (BO₃) units. Signals in the FTIR mainly observed around 1032 cm⁻¹, 1180 cm⁻¹, 1349 cm⁻¹, 1395 cm⁻¹ and 1501 cm⁻¹ while Raman scattering peaks are located at 436 cm⁻¹, 542 cm⁻¹, 770 cm⁻¹, 940 cm⁻¹, 1028 cm⁻¹, 1232 cm⁻¹ and 1361 cm⁻¹. Both reflective index and density increases with increasing rare earth ion concentration.

Keywords: Tellurium Glasses, FTIR Spectroscopy, Raman Scattering and Optical Absorption.

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Introduction

Since they are optical glasses containing Pr³⁺ ions, they are useful as photonic materials in various technologies due to their broad absorption in the visible-near infrared region [1]. Commercially available GaN diode lasers provide good utilization due to their broadness and absorption at 443 nm, which corresponds to the energy of the Pr³⁺ ion ³H₄ - ³P₂ [3]. Telluride networks consist mainly of TeO₃ trigonal pyramidal (Tp) units with long electron pairs and TeO₄ trigonal bipyramidal (Tbp) units. Add iron oxides, alkali metals and alkaline earth metals to convert Tbp units to Tp units. Raman spectroscopy and infrared (IR) spectroscopy are the two most important vibrational non-invasive tools in many fields such as biology, geology, materials science and recently space exploration [4]. Combined Raman-infrared spectroscopy can be used to analyze various important properties including phonons, aggregation state and energy of functional groups of the crystal lattice. Therefore, the properties of the material have a direct effect on the Raman spectroscopy parameters (band frequency, intensity, band shape and full width at half maximum or FWHM). These

constraints are: (1) Orientation of the crystal lattice or molecules; (5) Inconsistencies and stresses in the material (due to mechanical, thermal or acoustic effects) [5].

FTIR spectra recorded in the 1000–1600 cm⁻¹ region for Pr³⁺-doped alkali and mixed-alkali heavy metal borate glass matrices. In general, the vibrational modes of the borate glass network consist of three important band regions. The first group of bands, which occur at 1200–1600 cm⁻¹ is due to the asymmetric stretching vibration of B-O bonds in BO₃ units. Therefore, the aim of this study is the spectral and fluorescence analysis of Pr³⁺ telluride bismuth borate glass for optical devices.

Method

In the tellurite bismuth borate glass system 20TeO₂ was doped with Pr³⁺ ions as follows: (30-x) Using the solution quenching process, 20TeO₂: (30-x)Bi₂O₃: 50B₂O₃: XPr₂O₃ (where x = 0, 0.1, 0.3, 0.5 and 0.7 Mol%) was prepared. Chemical reagents, particularly TeO₂, Bi₂O₃, B₂O₃ and Pr₂O₃ were used in this study. Mix well using an agate mortar and pestle. After the lamb is melted in an electric

muffle furnace at 950 °C for two hours and completely melted, it is quickly poured into a preheated stainless-steel mold. The mold is placed at 350 °C for two hours to eliminate thermal stress and tension. These models are always polished with a fine cerium oxide powder.

Table 1: Chemical Composition of Pr³⁺ doped TEBIB Glasses.

Glass System	Glass Composition
TEBIB (Pr0)	20TeO ₂ :30Bi ₂ O ₃ :50B ₂ O ₃
TEBIB (Pr0.1)	20TeO ₂ :29.9Bi ₂ O ₃ :50B ₂ O ₃ :0.1Pr ₂ O ₃
TEBIB (Pr0.3)	20TeO ₂ :29.7Bi ₂ O ₃ :50B ₂ O ₃ :0.3Pr ₂ O ₃
TEBIB (Pr0.5)	20TeO ₂ :29.5Bi ₂ O ₃ :50B ₂ O ₃ :0.5Pr ₂ O ₃
TEBIB (Pr0.7)	20TeO ₂ :29.3Bi ₂ O ₃ :50B ₂ O ₃ :0.7Pr ₂ O ₃

Result and Discussion

All of the manufactured glasses were made of completely amorphous in nature, according to X-ray diffraction (XRD) patterns. From the figure 2 and 3 it is clear that reflective index and Density of the samples increase with increases in rare earth ion concentration respectively. From the FTIR spectra peaks are available in the range from 1000 cm⁻¹ to 1600 cm⁻¹ that reveals symmetrical stretching vibrations of Te-O in TeO₄ as well as B-O stretching (BO₃) units. The prominent feature in most synthetic tellurium borate glasses Raman spectra are intensive bands in the 400 – 1400 cm-region due to symmetric stretching motions of Te-O in TeO₄ as well as B-O stretching (BO₃) units.

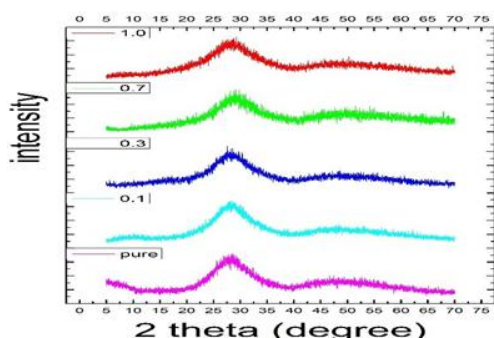


Figure1: X-ray diffraction pattern of Pr³⁺ doped TEBIB Glasses.

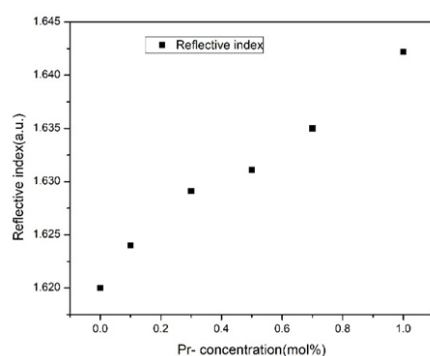


Figure 2: Reflective index v/s Pr- concentration (mol%) of Pr³⁺ doped TEBIB Glasses.

Table 2: Physical properties of Pr³⁺ doped TEBIB Glasses.

Properti es	TE BI B	TEBI B0.1P r	TEBI B0.3P r	TEBI B0.5P r	TEBI B0.7P r	TEBI B1.0P r
Reflecti ve index(n)	1.6200	1.6240	1.6291	1.6311	1.6350	1.6422
Density(gm/cm ³)	3.1021	3.2410	3.2641	3.3320	3.3785	3.4413

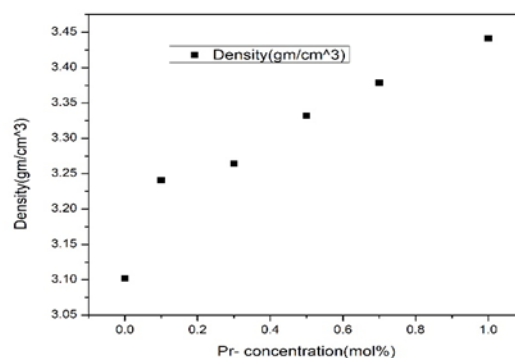


Figure 3: Density v/s Pr- concentration (mol%) of Pr³⁺ doped TEBIB Glasses.

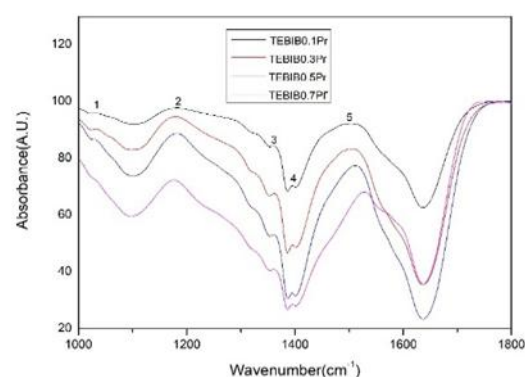


Figure 4: FTIR Spectra of Pr³⁺ doped TEBIB Glasses.

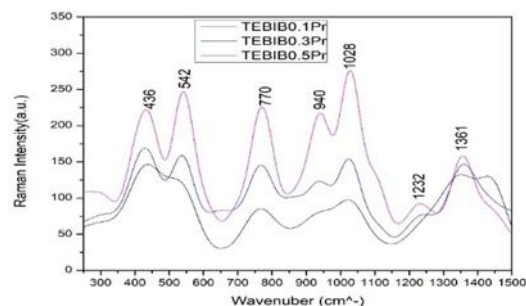


Figure 5: Raman Spectra of Pr³⁺ doped TEBIB Glasses.

Conclusion and Future Prospective

For a bismuth telluride borate glass sheet Pr₂O₃ with the composition of 20TeO₂:(30-X), undoped and Pr³⁺-doped Bi₂O₃:50B₂O₃:(X), use the usual melt quenching process to produce it (where X = 0, 0.1, 0.3, 0.5 and 0.7 mol%

vibration of Pr_2O_3 -doped tellurium). The glass was characterized by FTIR spectroscopy. FTIR shows the symmetric stretching vibrations of Te-O and B-O stretching (BO_3) units. Although Raman scattering peaks appear around 446 cm^{-1} , 542 cm^{-1} , 770 cm^{-1} , 940 cm^{-1} , 1028 cm^{-1} , 1232 cm^{-1} and 1361 cm^{-1} , the FTIR signal usually appears at 1032 cm^{-1} , 1180 cm^{-1} , 1349 cm^{-1} , 1395 cm^{-1} and 1501 cm^{-1} . Both reflective index and density increases with increasing rare earth ion concentration.

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